## STEREOCHEMISTRY OF A [2,3]SIGMATROPIC ALLYL-SULFOXIDE/ALLYL SULFENATE REARRANGEMENT

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(Received in UK 15 May 1978; accepted for publication 19 May 1978) [3,3]Sigmatropic rearrangements follow a well defined stereochemistry<sup>1)</sup> and lend themselves for asymmetric syntheses<sup>2)</sup>. Much less is known about the stereochemistry of [2,3]sigmatropic rearrangements<sup>3)</sup>, which also show high stereoselectivities<sup>4)</sup>, being even stereospecific in some cases<sup>5)</sup>. We were interested in applying the MISLOW/EVANS sequence<sup>6)</sup> to the asymmetric synthesis of alcohols by chirality transfer from sulfur to carbon and report here on the stereochemistry of the conversion of 2 via 3 to methylenecyclohexanol ( $\underline{4}$ ).



Starting from a single enantiomer of  $\underline{1}$  the above sequence led to  $\underline{4}$  of 60% ee<sup>\*</sup> in 75% yield<sup>7</sup>. Provided that  $k_2[(CH_3O)_3P]$  is >  $k_{-1}^{8}$  the stereochemistry of  $\underline{4}$  is defined in the thus irreversible step  $\underline{2} \rightarrow \underline{3}$ . Starting from two different rotamers of  $\underline{2}$  the [2,3]sigmatropic rearrangement can proceed via two diastereomeric transition states<sup>3</sup>, called "exo" and "endo" <sup>10</sup>. It is recognized that starting from the (R)-sulfoxide  $\underline{2}$  the endo transition state leads to (R)- $\underline{4}$ , whereas (S)- $\underline{4}$ 

<sup>\*)</sup> ee stands for enantiomeric excess and corresponds to optical purity.

is formed via the exo transition state. In order to assign the predominant path we have established the absolute configuration of  $\frac{4}{2}$  by chemical correlation of  $(+)-\frac{4}{2}$  with (1S,2S)-cyclohexanediol<sup>11)</sup> and of  $(-)-\frac{4}{2}$  with (1R,2RS)-methylcyclo-hexanol<sup>12)</sup>.



The obtention of dextrarotatory  $\frac{4}{2}$  from the (R)-sulfoxide  $\frac{1}{2}$  demonstrates the predominance of the exo transition state. This contrasts to the corresponding rearrangement of straight chain allyl sulfoxides via the endo transition state<sup>7)</sup>.

More insight into the stereochemistry of the  $2 \rightarrow 3$  rearrangement can be gained by dissecting the pathways leading to an equatorial and axial sulfenate. This has been accomplished with the tert.-butyl group as conformation anchor. A presumably 1:1 mixture of the diastereomers 5 was obtained from 4-tert.-butyl-cyclohexanone as described before<sup>7)</sup> and rearranged to the cis/trans-mixture of the 3-tert.-butyl-6-methylene-cyclohexanols  $\underline{7a}$  and  $\underline{8a}$  in 71% overall yield. The isomers were separated by vpc. Structural assignments rest on the following data: trans: <sup>1</sup>H-NMR CH-0 m, half width 6 Hz, <sup>13</sup>C-NMR C-3 40.4 ppm; =CH<sub>2</sub> 109.1; cis: <sup>1</sup>H-NMR CH-0 m, half width 19 Hz; <sup>13</sup>C-NMR C-3 47.0 ppm; =CH<sub>2</sub> 103.0.



The optical purities were determined from the Eu/TFC-shifted<sup>13)</sup> spectra. Since  $\underline{7b}$  and  $\underline{8b}$  are compounds of known absolute configuration<sup>14)</sup> the reaction sequence was repeated with an isopropyl-substituent. Yields, product ratios and signs of rotation <sup>15)</sup> of  $\underline{7b}$  and  $\underline{8b}$  were the same as in the tert.-butyl-series. We therefore assume that in both cases the reaction follows the same stereochemistry and that the absolute configuration of the pairs  $\underline{7a}$ ,  $\underline{7b}$  and  $\underline{8a}$ ,  $\underline{8b}$  are each the same. This then allows the following partial rate constants to be derived<sup>16)</sup>:



The [2,3] sigmatropic rearrangement of  $\underline{6}$  is apparently subject to two controlling factors: 1. a 6.2 fold preference for an axial versus an equatorial formation of the carbon-oxygen-bond. 2. a 3.1 fold preference for an exo versus and endo transition state. These effects enforce each other in the B-diastereomer and counteract in the A-isomer.

Applying the above partial rate constants to the rearrangement of  $\frac{2}{2}$  in which the A and B-isomers are conformers and not diastereomers, it may be surmised that the following reaction paths are used: exo axial: endo axial: exo equatorial: endo equatorial = 72:16:8:4. Thus the reaction via conformer B is 3 times favoured over a reaction via A. We gratefully acknowledge support of this study by the <u>Deutsche Forschungsge</u>meinschaft and the Fonds der Chemischen Industrie.

## References

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- 15) <u>7b</u>: 76 <sup>+</sup>/<sub>+</sub> 10% ee from EuTFC shifted <sup>1</sup><sub>1</sub>H-NMR spectra; 94 <sup>+</sup>/<sub>+</sub> 8% ee from rotation; <u>85</u>: 25 <sup>-</sup>/<sub>-</sub> 10% ee from EuTFC shifted <sup>1</sup>H-NMR spectra; 21 <sup>-</sup>/<sub>-</sub> 0.5% ee from rotation.
- 16) It is not known whether the 46:54 partitioning between the diastereomers <u>6aA</u> and <u>6aB</u> is significantly different from 50:50. In the <u>6b</u> series the value is 47:53. This might imply a slight stereoselectivity in the deprotonation of <u>5</u>